Design and Scalable Assembly of High Density Low Tortuosity Electrodes

P.I.: Yet-Ming Chiang

Presenter:

Massachusetts Institute of Technology, Cambridge, MA

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Project ID: ES071

Overview

Timeline

- Project start: May 22, 2013
- Project end: April 3, 2017
- 25% complete

Budget

- Total project funding
 - DOE share: \$1,075,344
 - Contractor share: \$0
- Funding received in FY13
 - \$256,983
- Funding for FY14
 - \$265,637

Barriers

- Lowering cost and increasing energy density of Li-ion batteries by reducing inactive content
- Achieve 3x times the area capacity (mAh/cm²) of current technology under EV duty cycles
- Achieving sufficient electronic conductivity in additive-free dense thick electrodes.

Partners

Collaborations: Antoni P. Tomsia, LBNL

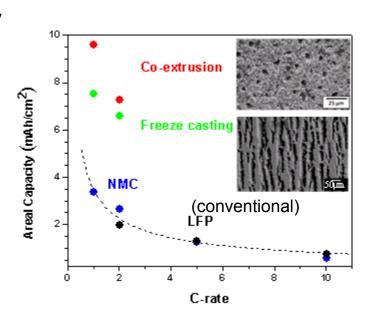
- Sample fabrication

Project lead: MIT

Relevance

- The high inactive materials content of current Li-ion batteries contributes directly to high battery cost, and reduces specific energy and energy density.
- The number of separator and current collector layers per Ah of cell capacity is inversely proportional to the area capacity (mAh/cm²) of the electrode.
- Area capacity can be increased arbitrarily by increasing electrode thickness and/or density, but does not contribute *usable* energy unless the capacity can be accessed at practical C-rates.
- Thus, concepts that can provide higher usable area capacity, e.g. during ~2C pulses in a EV or PHEV drive cycle, are needed.
- Binder-free, thick electrodes with low tortuosity porosity is one such approach, previously demonstrated in LiCoO₂ cathodes:

(Bae et al. *Adv.Mater*. 2013, *25*, 1254–1258, and present work.)



Objectives

Overall Objective: Develop a scalable high density binder-free low-tortuosity electrode design and fabrication process to enable increased cell-level energy density compared to conventional Li-ion technology.

Objectives in detail (March 2013 – March 2014):

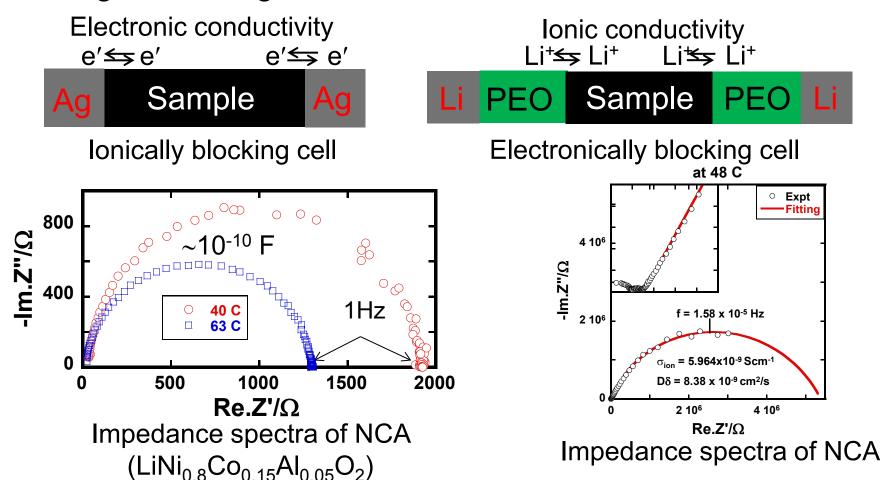
- Measure electronic and ionic transport in candidate cathode materials as a function of lithium content and temperature, using additive-free sintered dense samples to obtain results for the pure material.
- In addition to LiCoO₂ (previously validated) and NMC (previously ruled out for low electronic conductivity), candidates are LiNi_{0.8}Co_{0.15}Al_{0.05}O₂ (NCA), LiMn_{1.5}Ni_{0.5}O_{4-δ} (LMNO) and LiMn_{1.5}Ni_{0.5-y}Fe_yO_{4-δ} (Fe-LMNO).
- Down-select cathodes for further work based on electronic and ionic conductivity.
- Use directional freeze-casting to fabricate additive-free, dense electrodes with oriented low-tortuosity pore structure.
- Perform electrochemical testing of electrodes targeting area capacity of at least 5mAh/cm² under 1C continuous rate and 10mAh/cm² under 2C, 30 sec pulse.

Milestones

Quarter	Milestones/Deliverables Description and Due Date
Q1 10/1/13- 12/31/13	Measure electronic and ionic conductivities and diffusivity in sintered dense $Li(Ni,Co,Al)O_2$ (NCA) and Fe-doped high voltage spinel $Li_{1-x}Mn_{1.5}Ni_{0.5}O_4$. Fabricate first freeze-cast samples of at least one cathode composition from (12/31/13). Completed
Q2 1/1/14-3/31/14	Go/No-Go Down select one cathode composition for follow-on work based on transport measurements and cycling tests of freeze cast and sintered electrodes (3/31/14) Completed
Q3 4/1/14-6/30/14	Demonstrate at least 5mAh/cm²capacity per unit area at 1C continuous cycling rate for a freeze-cast cathode. (6/30/14)
Q4 7/1/14-9/30/14	Demonstrate at least 10mAh/cm²capacity per unit area for a 2C 30 sec pulse for a freeze-cast cathode. (9/30/14)

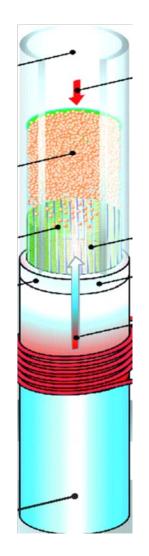
Approach

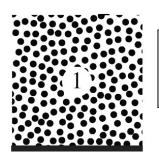
Independent measurement of the electronic and ionic conductivities by Electrochemical Impedance Spectroscopy (EIS) and Direct Current (DC) polarization techniques in a blocking cell configuration:



Approach

Freeze-Casted Electrode Fabrication:





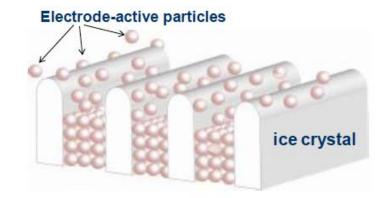
Slurry preparation

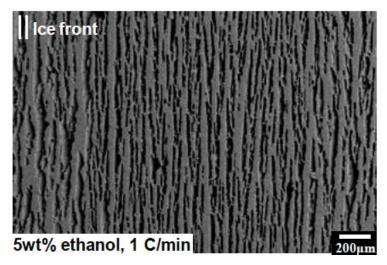


Solidification of the slurry



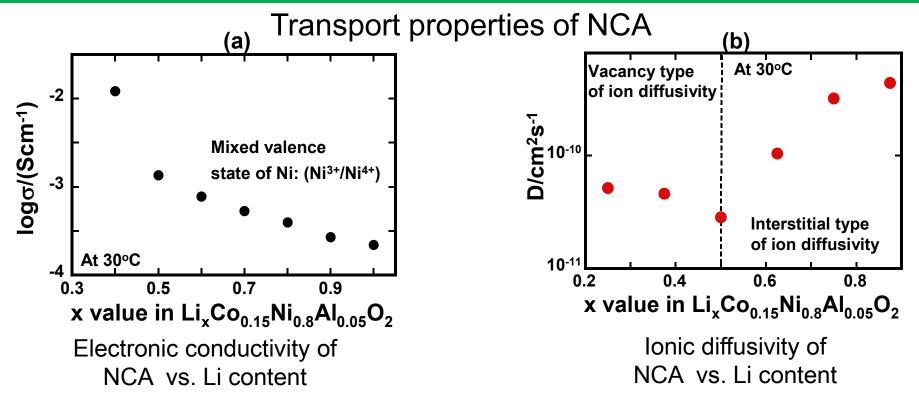
Sintering of the green body





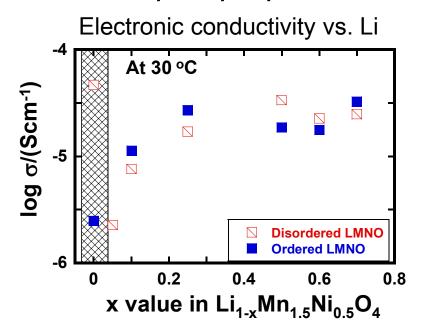
SEM micrograph of freeze casted LiCoO₂

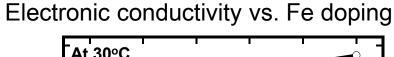
Schematic of freeze casting process

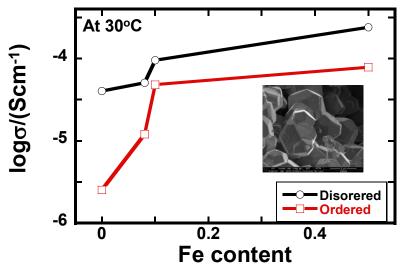


- (a) NCA has ~10⁻⁴–10⁻³ S/cm electronic conductivity over the Li concentration range of interest. The electronic conductivity of NCA increases upon delithiation, but is always lower than that of LiCoO₂.
- (b) It appears that lithium diffusion mechanism changes from interstitial type to vacancy type around 50% delithiation state. Obtained data can be functioned with the lamellae thickness up to 8µm for higher cycling rate.

The transport properties of LMNO and Fe-LMNO:

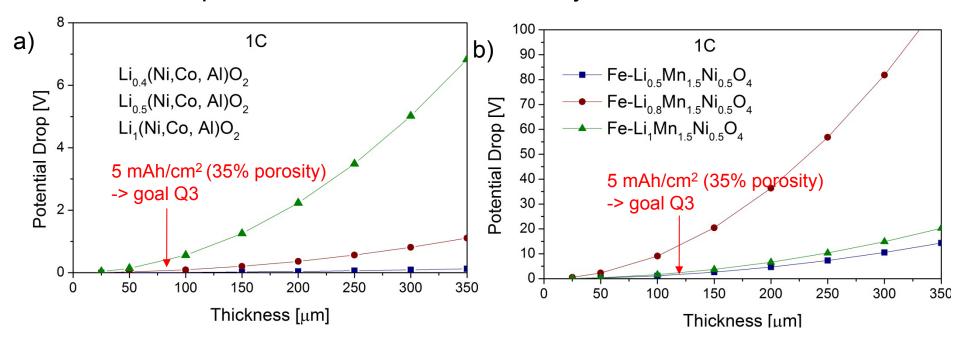






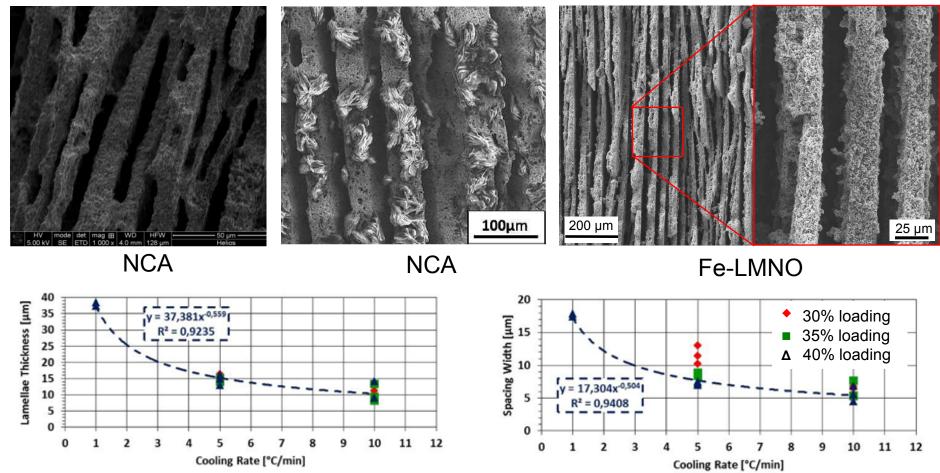
- Ionic Ionic Samples (at 50°C) conductivity diffusivity (cm²s⁻¹) (Scm⁻¹) ~10-7 DC LiMn_{1.5}Ni_{0.5}O₄ ~2×10⁻⁷ AC LiMn_{1.5}Ni_{0.5}O₄ $\sim 6 \times 10^{-8}$ ~5×10⁻⁸ ~8×10⁻⁸ ~6×10⁻⁸ DC $LiMn_{1.5}Ni_{0.42}Fe_{0.08}O_4$ ~6×10⁻⁸ ~9×10⁻⁸ $LiMn_{1.5}Ni_{0.42}Fe_{0.08}O_{4}$ AC
- Fe-LMNO selected for its resistance to electrochemical fracture
- Both LMNO and Fe-LMNO have a too low an electronic conductivity to use as a pure phase.
- Conductive additives would need to be incorporated for thick electrode applications.

Potential drop vs. electrode thickness for fully dense NCA and Fe-LMNO



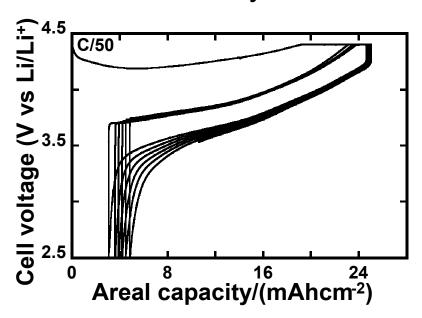
- Calculated for electronic conductivity at various Li contents. Note that cathodes in Li-ion cells are never fully lithiated after first cycle due to Li consumed in SEI formation
- Current density calculated on basis of measured capacities
- Potential drops are too large in Fe-doped LMNO to reach Q3 goal of 5 mAh/cm² at 1C rate – excluded from further consideration
- NCA to be evaluated further with electrochemical tests to establish true capacity vs. rate performance

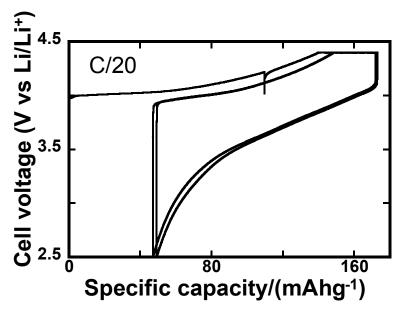
Freeze-cast and sintered NCA and Fe-LMNO structures



- The freeze casting process was successfully adopted to both cathode materials
- The pores are penetrating at least over 600 μm
- The lamellae thickness can currently be varied from 40 μm to 10 μm
- The spacing width can currently be varied from 17 μm to 5 μm

Preliminary tests of freeze-cast and sintered NCA





- These initial electrodes have~10 µm lamellae thickness, which is too large based on the solid phase ionic conductivity measurements
- At C/50, 24 mAh/cm² areal capacity is available from a 520 μm thick electrode of 60-70% density (specific capacity is 160 mAh/g.)
- At C/20 the same electrode has 120mAh/g reversible capacity, and there is a clear increase in polarization
- Next experiments will focus on samples with smaller lamellae thickness

Response to previous year Reviewer's Comments

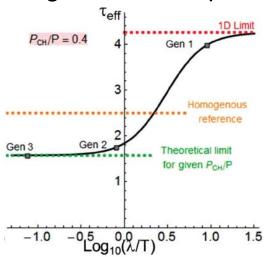
Not applicable – first year of project.

Collaboration/Coordination with Other Institutions

- Collaboration with A.P. Tomsia group at Lawrence Berkley National Laboratory (LBNL) for fabrication of directionally freeze-cast electrodes
- PI is member of BATT's Silicon Focus Group
- PI is member of BATT's High Voltage Cathode Focus Group and is Team Lead for Cathode Crystal Structure Transformations subgroup
- Dr. Jonathan Sander from ETH Zurich is self-supported member of BATT project working on alternative methods for pore alignment

Remaining Challenges and Barriers

 Fabrication of thick dense electrodes with controlled dual-scale porosity: aligned pore channels having wall thickness below 8µm and a higher degree of micro porosity in the walls.



Previous calculations show that shorter channel spacing (λ) for a given thickness T more effectively reduces the tortuosity and thus ion diffusion in the electrolyte.

Calculations also suggest that less than about 50% of the pore volume should be in the channels P_{CH} (Bae et al. Adv.Mater. 2013, 25, 1254–1258)

 Developing a detailed model for transport across such electrodes, including continuous and pulse response.

Proposed Future Work

Rest of FY 2014

- Tailor the microstructure in NCA in pursuit of the Milestones and Objectives of the project. Prepare and characterize comparative sintered porous samples without aligned pores. Revisit LiCoO₂.
- Demonstrate at least 5mAh/cm²capacity per unit area at 1C continuous cycling rate for a freeze-cast cathode.
- Demonstrate at least 10mAh/cm²capacity per unit area for a 2C 30 sec pulse for a freeze-cast cathode.

FY 2015

- Initial development of directional freeze-casting in a planar geometry.
- Develop model to predict the impact of electronic conductivity on the ohmic potential drop in a porous electrode.
- Test best available cathodes in half-cells using USABC PHEV or EV protocols.
- Develop first additive-free anodes for use with microstructually tailored cathodes.
- Incoporate Si anodes from the BATT Si Focus Group
- Construct and test full Li-ion cells

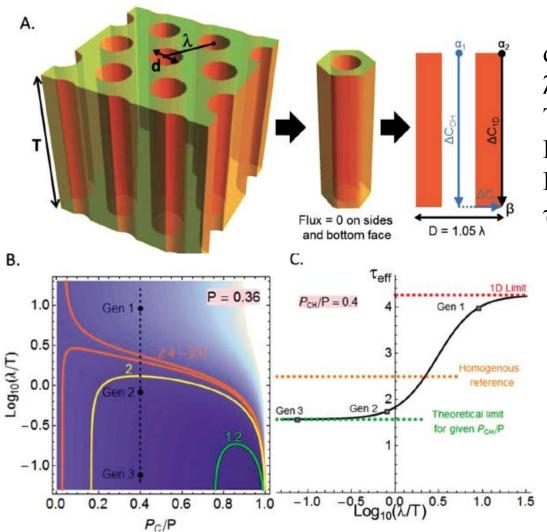
Presentation Summary

- The electronic conductivity of the cathode is a critical parameter to obtain high area capacity (>10 mAh/cm²) with dense additive-free electrodes. During the current fiscal year, the electronic and ionic conductivity of NCA, LMNO and Fe-LMNO (which is electrochemical-shock resistant) have been characterized.
- Of all the cathodes considered, NMC, LMNO and Fe-LMNO have been determined to have too low an electronic conductivity for use as additive-free electrodes. NCA and LCO (previously tested) have potential and will be further evaluated.
- Electrode structures with aligned pore channels have been successfully produced in NCA, LMNO and Fe-LMNO by directionalfreeze casting. NMC and LiCoO2 were previously demonstrated, and illustrate the generality of the approach
- The microstructure of the freeze-casted samples can be systematically varied (i.e. channel spacing and width) by controlling freeze-casting rate and sintering conditions

Technical Back up slides

Technical backup slide

Influence of channel spacing and channel pore fraction on tortuosity



d...channel diameter

λ... channel spacing

T... electrode thickness

P_{CH}... pore fraction in channels

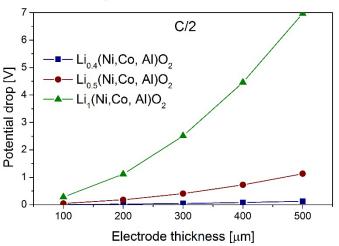
P... total pore fraction

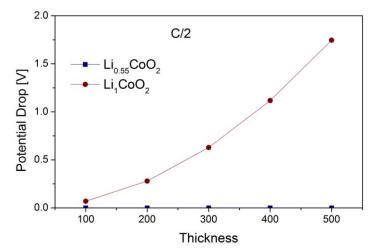
τ... tortuosity

Technical backup slide

Estimation of Ohmic drop

$$\Delta V = RI = t^2 \frac{1}{\sigma} \rho (1 - p) * Crate * Cap$$





 ΔV ... voltage drop, R... resistivity, I... current [Ah], t..., thickness, ρ ... density $[g/cm^3]$, A... area: $1cm^2$, p... porosity, Crate [1/h], Cap... capacity [Ah/g], σ ... conductivity,

